Table 3. Hydrogen bonds in hesperetin monohydrate

$a-b\cdots c$	<i>b</i> – <i>c</i> (Å)	<i>a−c</i> (Å)	abc (°)
O(5)-H····O(4)	1.67 (6)	2.552 (5)	137 (5)
$O(7)-H\cdots O(W^{1})$	1.68 (5)	2.713 (7)	170 (4)
O(3')–H···O(5 ⁱⁱ)	1.64 (7)	2.731 (4)	174 (5)
$O(W) - H(1) \cdots O(3'^{iii})$	2.17 (6)	2.795 (7)	170 (7)
$O(W) H(2) \cdots O(4'^{iv})$	2.47 (10)	3.159 (7)	175 (10)
O(5)–H···O(4' ^{iv})	2.46 (6)	2.907 (4)	104 (4)
O(3')–H···O(4')	2.36 (6)	2.678 (5)	95 (4)

Symmetry code: none x, y, z; (i) x, y, 1+z; (ii) x, 1+y, z; (iii) 1-x, 1-y, 1-z; (iv) x, -1+y, z.

only van der Waals interactions between the molecular double layers.

We thank the Korea Research Foundation for their support of this research.

References

- CAILLET, J. (1982). Acta Cryst. B38, 1786-1791.
- GILMORE, C. J. (1984). J. Appl. Cryst. 17, 42-46.
- GLUSKER, J. P. & ROSSI, M. (1986). Plant Flavonoids in Biology and Medicine: Biochemical, Pharmacological, and Structure– Activity Relationships, pp. 395–410. New York: A. R. Liss.

International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)

- JOHNSON, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- NYBURG, S. C. (1974). Acta Cryst. B30, 251-253.
- Rossi, M., Rickles, L. F. & Halpin, W. A. (1986). *Bioorg. Chem.* 14, 55–69.
- SHELDRICK, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England. SHIN, W. & LAH, M. S. (1986). Acta Cryst. C42, 626-628.
- Уамато, М., Назнідакі, К., Міто, К. & Коуама, Т. (1978).

Acta Cryst. (1987). C43, 1949–1951

Structure of 11,12-Dihydro-6H-6,12-methanodibenzo[b,f]thiocin-12-carbonitrile

By J. Podlaha and J. Podlahová

Department of Chemistry, Charles University, 128 40 Prague, Czechoslovakia

and J. Symerský

Institute of Organic Chemistry and Biochemistry, Czechoslovak Academy of Sciences, 166 10 Prague, Czechoslovakia

(Received 19 January 1987; accepted 18 May 1987)

Abstract. $C_{17}H_{13}NS$, $M_r = 263 \cdot 36$, $P2_1/c$, $a = 8 \cdot 146$ (3), $b = 18 \cdot 660$ (8), $c = 8 \cdot 666$ (2) Å, $\beta = 91 \cdot 41$ (3)°, $V = 1316 \cdot 9$ (8) Å³, Z = 4, $D_m = 1 \cdot 296$ (5), $D_x = 1 \cdot 33$ Mg m⁻³, λ (Mo K α) = 0.71069 Å, $\mu = 0.213$ mm⁻¹, F(000) = 552, T = 295 K, R = 0.0425 for 1836 observed unique reflections. The molecular structure is tetracyclic with a central bicyclo[3.3.1]-2-thianonane-3,7-diene moiety and two phenyl rings ortho-condensed to it. Bond lengths and angles are normal.

Introduction. The phase-transfer-catalysed reaction of 6,11-dihydrodibenzo[*b,e*]thiepin-11-carbonitrile with excess 1,2-dibromoethane (Šindelář, Holubek, Ryska, Svátek, Urban & Protiva, 1983) afforded three isomeric products of composition $C_{17}H_{13}NS$. On the basis of NMR and mass spectra, the structure below was tentatively assigned to one of the minor products (compound XVI in the original paper). Subsequent detailed ¹H and ¹³C NMR data (Šindelář *et al.*, 1987) indicated, however, that the structure should not involve a cyclopropane ring. The X-ray crystal struc-

ture determination has been undertaken to resolve the ambiguities.



Experimental. Single crystals grown from benzene/ cyclohexane were kindly provided by K. Šindelář. The density was determined by flotation in aqueous zinc bromide solution. The measurement and refinement data are summarized in Table 1.*

Structure solution and refinement: direct methods; F magnitudes used in full-matrix least-squares refinement; all H atoms found from $\Delta \rho$ synthesis; scale

© 1987 International Union of Crystallography

Chem. Pharm. Bull. 26, 2321-2327.

^{*} Lists of structure factors, anisotropic thermal parameters and H-atom coordinates have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 44060 (11 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

factor, positions and anisotropic thermal parameters of non-H atoms, positions of aliphatic H atoms, a common isotropic thermal parameter for aliphatic H atoms and another common thermal parameter for aromatic H atoms refined simultaneously; aromatic H atoms fixed in calculated positions; programs used: MULTAN80 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980), SHELX76 (Sheldrick, 1976); ICL 4-72 computer. Atomic scattering factors those of SHELX76.

atom numbering. (The numbering is somewhat unusual from the point of view of crystallographic conventions but follows the IUPAC nomenclature rules.) Bond lengths and angles are summarized in Table 3.

The structure consists of discrete molecules packed (Fig. 2) at van der Waals or longer distances. In

Table 3. Bond lengths (Å) and angles (°) with e.s.d.'s in parentheses

1.378(4)

1.369(4)

1.394 (4)

1.402 (3)

1.392 (3)

1.390 (3)

1.371 (4)

1.391 (4)

1.376 (4)

1.400 (3)

1.386 (3)

1.397 (3)

1.764 (2)

1.826 (6)

1.513(3)

1.544 (3)

1.541 (3) 1.507 (3)

1.521 (4)

1.536 (3)

1.488 (4)

1.128(3)

C(12A) - C(1)

C(7)-C(8)

C(8) - C(9)

C(9)-C(10)

C(1)-C(2)-C(3) C(2)-C(3)-C(4) C(3)-C(4)-C(4A)

C(4) - C(4A) - C(12A)

C(4A) - C(12A) - C(1)

C(8)-C(9)-C(10) C(9)-C(10)-C(10A) C(10)-C(10A)-C(6A)

C(10A) - C(6A) - C(7)

C(6A)-C(7)-C(8)

C(4)-C(4A)-S C(12A)-C(4A)-SC(4A)-S-C(6)

S-C(6)-C(6A)

C(6) - C(6A) - C(7)

C(6) - C(6A) - C(10A)

C(4A) - C(12A) - C(12)

C(1)-C(12A)-C(12)C(12A)-C(12)-C(11)

C(12)-C(11)-C(10A)C(11)-C(10A)-C(6A)

C(11)-C(10A)-C(10)

C(6)-C(13)-C(12)

C(6A) - C(6) - C(13)

C(12A) - C(12) - C(13)

C(11)-C(12)-C(13)

C(11)-C(12)-C(14)

C(12A)-C(12)-C(14) C(13)-C(12)-C(14) C(12)-C(14)-N

S-C(6)-C(13)

C(12A) - C(1) - C(2)

C(7)-C(8)-C(9)

120-1 (3)

119.7 (2)

120.4 (3)

119.4 (2)

118.9 (2)

121.5 (3)

119-5 (3)

120-1 (3)

120.7 (3)

119.2 (2)

119.4 (2)

121.1 (3) 115-7 (2)

124·9 (2) 99·7 (1)

111.2(2)

119-8 (2)

120-8 (2)

123-2 (2)

117.9 (2)

110.6 (2)

112.2(2)122.2 (2)

118.6 (2)

108-9 (2)

114.1 (2)

107.9 (2)

112.9 (2)

108.3 (2)

107.0 (2)

109.7 (2) 108-1 (2)

176-4 (3)

C(1)-C(2) C(2) - C(3)Discussion. The atomic coordinates are given in Table C(3) - C(4)2; Fig. 1 depicts a perspective view of the molecule with C(4) - C(4A)C(4A) - C(12A)

Table 1. Measurement and refinement data

Crystal dimensions (mm)	$0.2 \times 0.3 \times 0.15$	C(10) - C(10A)
No. of reflections for lattice- parameter determination	20 (4 < 2θ < 25°)	C(10A)-C(6A) C(6A)-C(7)
Diffractometer	Syntex P2,	
Scan mode	$\theta - 2\theta$	C(4A) - S
Maximum value of $(\sin\theta)/\lambda(\dot{A}^{-1})$	0.60	S = C(6) C(6) = C(6.4)
Standard reflections (variation)	Three after every 30 (less than 3%)	C(0)-C(0A)
Absorption correction	None, $\mu = 0.213 \text{ mm}^{-1}$	
h∈	(0,9)	
k∈	(0,22)	C(12A) - C(12)
l∈	(-10,10)	C(12) - C(11)
No. of unique reflections measured	2512	C(11) - C(10A)
No. of reflections used $[I > 1.96 \sigma_1(I)]$	1836	
Residual electron density (e Å-3)	0.20, -0.23	C(6)-C(13)
$(\Delta/\sigma)_{\rm max}$ for non-H atoms	0.03	C(13)C(12)
for H-atoms	0.15	
Function minimized	$w(F_{o} - F_{c})^{2}$	
Weight	$\sigma^{-2}(F_{\alpha})^*$	
R, wR	0.0425, 0.0501	C(12)–C(14) C(14)–N

* Based on $\sigma_1(I) = \text{scan rate} \times (\text{total scan counts} + \text{left back})$ ground + right background)^{1/2}; $\sigma_2(I) = [\sigma_1^2(I) + (0.06 I)^2]^{1/2};$ $\sigma(F_o)/F_o = \frac{1}{2} \sigma_2(I)/I.$

parentheses							
isotrop	vic	thermai	parameters	(× 10 ⁴)	with	e.s.d.'s	in
Table	2.	Final	coordinates	(×10 ⁴)	and	equivale	ent

 $U = \frac{1}{2} (U + U + U + 2U \cos \theta)$

$U_{eq} = \frac{1}{3}(U_{11} + U_{22} + U_{33} + 2U_{13} + U_{13})$					
	x	у	z	$U_{eq}(\text{\AA}^2)$	
S	2535 (1)	423-3 (4)	467 (1)	616 (3)	
N	-233 (4)	2868 (1)	2971 (4)	938 (12)	
C(1)	3953 (3)	2311 (1)	2426 (3)	384 (8)	
C(2)	5462 (3)	2412 (2)	1776 (3)	517 (10)	
C(3)	6025 (3)	1933 (2)	715 (3)	672 (10)	
C(4)	5081 (3)	1334 (2)	322 (3)	598 (9)	
C(4A)	3557 (3)	1222 (1)	1001 (2)	449 (7)	
C(6)	1036 (3)	378 (1)	1995 (3)	531 (8)	
C(6A)	1834 (3)	147 (1)	3509 (3)	457 (7)	
C(7)	2307 (3)	-567 (1)	3721 (3)	615 (10)	
C(8)	3091 (4)	-785 (2)	5058 (4)	696 (11)	
C(9)	3398 (3)	-292 (2)	6233 (3)	697 (11)	
C(10)	2922 (3)	411 (1)	6049 (3)	560 (9)	
C(10A)	2135 (3)	639 (1)	4681 (2)	425 (7)	
C(11)	1604 (3)	1411 (1)	4551 (3)	499 (8)	
C(12)	1322 (3)	1644 (1)	2860 (3)	460 (7)	
C(12A)	2977 (3)	1724 (1)	2044 (2)	424 (7)	
C(13)	169 (3)	1099 (2)	2065 (3)	576 (9)	
C(14)	467 (3)	2348 (2)	2890 (3)	642 (10)	



Fig. 1. Perspective view of the molecule (40% probability).



Fig. 2. Unit-cell contents projected onto the bc plane.

Table 4. Comparison of endocyclic torsion angles (°, e.s.d.'s in parentheses) of bicyclo 3.3.1 nonadiene systems

A,B: the present structure; C: Radcliffe et al. (1984); D: Quast et al. (1984) (mean values of four closely related structures); E: theoretical values for the ideal half-chair.

A			В	С	D	Ε
C(4A)C(12A)	-0.7 (4)	C(6A) - C(10A)	3.6 (4)	0.2(3)	- 1(1)	0
C(12A) - C(12)	-17.6 (3)	C(10A) - C(11)	$-21 \cdot 3(4)$	$-16 \cdot 1 (4)$	-13 (3)	15
C(12) - C(13)	57.4 (4)	C(11) - C(12)	51-5 (3)	50-5 (3)	47 (3)	46
C(13)-C(6)	-76.0 (3)	C(12)-C(13)	-65.5(4)	-68.7(3)	-68(2)	63
C(6)-S	50.8 (3)	C(13)-C(6)	48.1 (3)	51.0 (3)	52 (1)	46
S-C(4A)	-15.0 (3)	C(6)-C(6A)	-17.3 (4)	-16-7 (4)	18 (2)	- 15

accordance with recent NMR data (Šindelář et al., 1987), the molecular skeleton is tetracyclic with two phenyl rings ortho-condensed to a central bicyclo-[3.3.1]-2-thianonane-3,7-diene system. The bond lengths and angles are all normal, indicating an essentially strain-free conformation. Idealized geometry calculations show that there are two such conformations available which differ in the puckering of the central bicyclononadiene system and, consequently, in the dihedral angle of the aromatic ring planes: (a)twin-sofa central rings with dihedral angle ca 140°, and (b) twin-half-chair central rings with dihedral angle ca 90°. In the crystal, the (b) conformation is realized: the dihedral angle has the value of $91.2(3)^{\circ}$ and the idealized twofold half-chair axes bisect the C(4A)-C(12A) and C(6A)-C(10A) bonds respectively. This arrangement is closely related (Table 4) to the bicyclo[3.3.1]nonadiene systems whose crystal structures are known (Radcliffe, Gutierrez, Blount & Mislow, 1984; Quast, Gorlach, Stawitz, Peters, Peters & von Schmering, 1984). To our knowledge, no example of such a bicyclic system involving sulfur as a heteroatom has been characterized by crystal structure

determination till now. The asymmetry parameters (Duax, Weeks & Rohrer, 1976) $\Delta C_2(4A-12A) = 5 \cdot 0^{\circ}$, $\Delta C_2(6A-10A) = 3 \cdot 5^{\circ}$ indicate, however, that the S atom affects the relative conformation of the rings very little.

References

- DUAX, W. L., WEEKS, C. M. & ROHRER, D. C. (1976). Top. Stereochem. 9, 271-383.
- MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J.-P. & WOOLFSON, M. M. (1980). MULTAN80. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
- QUAST, H., GORLACH, Y., STAWITZ, J., PETERS, E.-M., PETERS, K. & VON SCHMERING, H. G. (1984). Chem. Ber. 117, 2745–2760.
- RADCLIFFE, M. D., GUTIERREZ, A., BLOUNT, J. F. & MISLOW, K. (1984). J. Am. Chem. Soc. 106, 682–687.
- SHELDRICK, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England.

ŠINDELAŘ, K., BUDĚŠINSKÝ, M., VANĚK, T., HOLUBEK, J., SVÁTEK, E., MATOUŠOVÁ, O., REES, C. W. & PROTIVA, M. (1987). Coll. Czech. Chem. Commun. In the press.

ŠINDELÁŘ, K., HOLUBEK, J., RYSKA, M., SVÁTEK, E., URBAN, J. & PROTIVA, M. (1983). Coll. Czech. Chem. Commun. 48, 1898– 1909.

Acta Cryst. (1987). C43, 1951–1954

The Structure of the 1:1 Complex of β -1',2',3',4',10,11-Hexahydrocinchonine and β -9-Deoxy-1',2',3',4',10,11-hexahydrocinchonine

By M. GDANIEC AND Z. KOSTURKIEWICZ*

Department of Crystallography, Faculty of Chemistry, A. Mickiewicz University, 60-780 Poznań, Poland

AND B. GOLANKIEWICZ

Institute of Bioorganic Chemistry, Polish Academy of Science, 61–704 Poznań, Poland

(Received 10 November 1986; accepted 13 April 1987)

Abstract. $C_{19}H_{28}N_2O.C_{19}H_{28}N_2$, $M_r = 584.9$, orthorhombic, $P2_12_12_1$, a = 12.463 (2), b = 13.227 (3), c

* To whom correspondence should be addressed.

= 20.330 (4) Å, V = 3351 (1) Å³, Z = 4, $D_m = 1.13$, $D_x = 1.16$ Mg m⁻³, λ (Cu Ka) = 1.54178 Å, μ (Cu Ka) = 0.46 mm⁻¹, F(000) = 1280, room temperature, R= 0.047 for 2297 observed reflexions. The complex-

0108-2701/87/101951-04\$01.50 © 1987 International Union of Crystallography